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The central atom Cl needs three unpaired electrons to bond with three F-atoms. ClF₃ should consist of 3 bond-pairs and 2 lone-pairs. One 3s, three 3p and one of the 3d orbitals of Cl participate in the hybridization and five sp³ d hybrid orbitals are formed.

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Bond enthalpy is an amount of energy required to break one mole of the stated bond. Learn Definition, Endothermic and Exothermic Process and Average bond enthalpies and more.

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Example: Calculate number of protons, electrons and neutrons for 17 Cl 35 or 35 17 Cl Since Cl is neutral, No. of electrons = no. of protons = 17 Mass no. of Cl = 35

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