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in Schroeder),  $\Delta H_{H_2O} = 285.83 \text{ kJ mol}^{-1}$  and  $\Delta H_{C_3H_8} = 890.36$

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$\text{kJ mol}^{-1}$  (12) This time, the final volume is  $\frac{1}{3}$  of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work  $P\Delta V = RT\Delta n$  (13) =  $8.31 \text{ J K}^{-1} (298 \text{ K})(2)$  (14) =  $4.953 \text{ kJ}$  (15)

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